Cambridge Assessment International Education

Cambridge IGCSE™

CO-ORDINATED SCIENCES **0654/52**

Paper 5 Practical Test **February/March 2023**

MARK SCHEME

Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the February/March 2023 series for most Cambridge IGCSE™, Cambridge International A and AS Level components and some Cambridge O Level components.

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Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5: Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).
GENERIC MARKING PRINCIPLE 6: Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	<u>'List rule' guidance</u> For questions that require ***n*** responses (e.g. State **two** reasons …): • The response should be read as continuous prose, even when numbered answer spaces are provided. • Any response marked *ignore* in the mark scheme should not count towards ***n***. • Incorrect responses should not be awarded credit but will still count towards ***n***. • Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response. • Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Acronyms and shorthand in the mark scheme

acronym/shorthand	**explanation**
AW	Alternative wording
Brackets ( )	Words not explicitly needed in an answer, however if a contradictory word/phrase/unit to that in the brackets is seen the mark is not awarded.
Underlining	The underlined word (or a synonym) must be present for the mark to be scored. If the word is a technical scientific term, the word must be there.
/ or **OR**	Alternative answers any one of which gains the credit for that mark.
owtte	Or words to that effect.
ecf [question part]	Indicates that a candidate using an erroneous value from the stated question part must be given credit here if the erroneous value is used correctly here.
cao	correct answer only

<table>
<tr><th>Question</th><th>Answer</th><th>Marks</th></tr>
<tr><td>1(a)</td><td>
A – red / purple / brown;
B – blue / purple / blue-black;

<table>
<tr><th>solution</th><th>final colour observed with Benedict’s solution</th><th>final colour observed with biuret solution</th><th>final colour observed with iodine solution</th></tr>
<tr><td>A</td><td>red;</td><td>purple;</td><td>brown</td></tr>
<tr><td>B</td><td>blue</td><td>purple;</td><td>blue-black;</td></tr>
</table>
</td><td>5</td></tr>
<tr><td>1(b)</td><td>both contain protein;
A also contains reducing sugar
B also contains starch;</td><td>3</td></tr>
</table>

Question	Answer	Marks
2(a)	headings for fruit juice and number drops; minimum of cross shape to separate headings and results;	**2**
2(b)	set of results for fruit juice **E**; full set of results (for **F**, **G** and **H**); **F** uses most drops / **H** uses least drops; all four in the right order;	**4**
2(c)	drop size varies; measure volume;	**2**
2(d)(i)	**H** because it requires fewest drops / highest vitamin C concentration;	**1**
2(d)(ii)	125 divided by 2 or 62.5 (cm^3); answer to 2sf (63);	**2**

Question	Answer	Marks
2(d)(iii)	other juices / food contains vitamin C;	1

Question	Answer	Marks
3(a)	temperature of sodium hydroxide ; all other temperatures recorded (5 cm^3–45 cm^3); temperatures increase then decrease ; all temperatures to 0.5 °C ;	4
3(b)(i)	axes correct way round and labelled with quantity and unit ; sensible linear axes with plotted points ± 10 °C covering ⩾ half of the grid ; points plotted correctly ;	3
3(b)(ii)	best fit line increasing temperatures ; best fit line decreasing temperatures ; extrapolated to cross ; temperature at intersection ± ½ small square and working on the graph;	4
3(b)(iii)	V from graph ± ½ small square;	1
3(c)	both calculations correct ;	1
3(d)	put lid on plastic cup / insulate the beaker/cup;	1
3(e)	room temp. / temp of initial NaOH quoted AND cools to room temp;	1

Question	Answer	Marks
4(a)	red **;** blue ; pH 0, 1, 2, 3 AND pH between 10 and 14 ;	**3**
4(b)(i)	red **and** pH between 1 and 3 ;	**1**
4(b)(ii)	more acid added after reaction finished OWTTE ;	**1**

Question	Answer	Marks
5(a)	correct symbol ; correct parallel connection in correct place;	**2**
5(b)(i)	*V* recorded and < 3 V; *I* recorded and < 1 A;	**2**
5(b)(ii)	R_S calculation correct; Ω / ohms / ohm;	**2**
5(c)(i)	*V* and *I* recorded; *I* > the value in **(b)(i)**;	**2**
5(c)(ii)	R_P calculation correct;	**1**
5(d)	so that cell does not run down / resistors / circuit / wires do not overheat	**1**

Question	Answer	Marks
5(e)	attempt at calculation of 4 times the value of R_P ; correct matching statement (yes / no – expect yes) and suitable comment e.g. values are too different / not close enough / >10% different;	**2**
5(f)	inspect the filament of each lamp / check if the ammeter gives a reading;	**1**

Question	Answer	Marks
6	**additional apparatus:** stop-watch ; thermometer ; **method:** pour (cold) water into one of the pots, boil it, measure time and repeat for all 3 pots; repeats method for each pot (i.e. more than one trial for each pot) **control variables:** mass / volume of cold water ; same initial cold-water temperature ; keep Bunsen burner flame constant ; **table:** columns for pot / metal and time ; units for time ; **processing and conclusion:** plot bar chart / histogram of metal against time; if the times are different / same then metal makes a difference / does not make a difference / lowest time best heat conductor ; calculate average identifying / excluding anomalies ;	**7**